



## ASSESSMENT OF NO<sub>x</sub> EMISSIONS FROM NITRIC ACID PRODUCTION AND THEIR EFFECT UPON AMBIENT AIR QUALITY IN DEVNYA REGION

**Rozalina Chuturkova, Maya Stefanova**

TECHNICAL UNIVERSITY VARNA  
DEPARTMENT OF ECOLOGY AND ENVIRONMENTAL PROTECTION  
1 STUDENTSKA STR., 9010 VARNA, BULGARIA  
E-MAIL: [chuturkova@hotmail.com](mailto:chuturkova@hotmail.com)

**Abstract:** Within this research data from monitoring of ambient air quality in Devnya region for the period 2006 – 2012 regarding the concentration of oxides of nitrogen NO<sub>x</sub> has been analyzed along with data from monitoring of NO<sub>x</sub> and nitrous oxide N<sub>2</sub>O emissions from a nitric acid plant situated in the industrial zone of Devnya for the same period. Results indicate that NO<sub>x</sub> and N<sub>2</sub>O emissions have significant effect upon the quality of the atmospheric air in the region but monitoring data is not a sufficient source to assess the contribution of these pollutants to ambient air quality. A necessity of mathematical modeling the diffusion of N<sub>2</sub>O emissions is concluded having in mind the processes of photo dissociation, chemical destruction and dispersion of this air pollutant that occur in the ground atmospheric layer and estimating the effect of the secondary pollutants proceeded from these reactions upon ambient air quality.

**Key words:** ambient air quality, greenhouse gas, mathematical modeling, nitrous oxide emissions, oxides of nitrogen

### I. Introduction

Nitric acid production is a major industrial source emitting oxides of nitrogen NO<sub>x</sub> that contain mainly nitric oxide NO, nitrogen dioxide NO<sub>2</sub> and nitrous oxide N<sub>2</sub>O. Nitrous oxide is a greenhouse gas under the Kyoto Protocol [1] and one of the main reasons for global warming effect [2]. Besides causing adverse changes of the climate system this greenhouse gas emissions have significant effect upon ambient air quality in the regions where industrial sources of N<sub>2</sub>O are available. In order to protect human health and the environment as a whole it is extremely important to

combat harmful emissions at source. Contributions from natural sources can be assessed but can not be controlled that's why it is a matter of great importance to apply most effective emission abatement techniques at emission source especially when it comes to industrial sources [3-5]. Various techniques for N<sub>2</sub>O emission reduction at nitric acid production are developed worldwide [6-8]. Regionally reducing N<sub>2</sub>O emissions from an industrial source is proved to have positive effect upon ambient air quality [11-12] while globally it is a tool for global warming prevention.

The present research aims to assess the effect of NO<sub>x</sub> emissions from an industrial source situated in the industrial zone of Devnya upon ambient air quality in the region.

## **II. Material and methods**

The research is done in the industrial zone of Devnya, Bulgaria where an industrial emission source is situated – emission source P1 at a nitric acid plant. Within this research data from an automatic system for continuous monitoring of NO<sub>x</sub> and N<sub>2</sub>O concentration in the tail gas from emission source P1 for the period 2006 – 2012 has been analyzed, data being expressed as average monthly concentrations and average annual concentrations in mg/m<sup>3</sup>. Regarding the assessment of ambient air quality in Devnya region monitoring data about the average monthly concentrations and average annual concentrations of NO<sub>x</sub> in µg/m<sup>3</sup> for the same period has been used – monitoring data is collected by the automatic monitoring system “Izvorite” which is a part of the National monitoring system for ambient air quality at the Ministry of Environment and Water. The monitoring system “Izvorite” is not equipped to measure the concentration of N<sub>2</sub>O in the ambient air and for that reason it is possible to observe only the tendency for NO<sub>x</sub> pollution. The monitoring period is coordinated with the initial implementation of N<sub>2</sub>O emission abatement technique at the nitric acid plant [11] by installing a secondary decomposing catalyst and an

automatic system for continuous monitoring of NO<sub>x</sub> and N<sub>2</sub>O concentrations in the tail gas in 2006.

Concentration of NO<sub>x</sub> in the ambient air is the summary concentration of NO and NO<sub>2</sub> added as parts per billion (ppb) and expressed as concentration of NO<sub>2</sub> in µg/m<sup>3</sup> [9]. Emission limit values ELV for NO<sub>x</sub> concentrations refer to summary emissions of NO and NO<sub>2</sub> expressed as NO<sub>2</sub> in mg/m<sup>3</sup> [10]. All ELV and air pollutant concentrations in µg/m<sup>3</sup> or mg/m<sup>3</sup> refer to temperature 273 K and pressure 1013 hPa including after humidity correction [10].

## **III. Results and discussion**

### **III.1. Quantitative assessment of NO<sub>x</sub> emissions from nitric acid plant**

Monitoring data regarding average monthly NO<sub>x</sub> concentrations (mg/m<sup>3</sup>) in the tail gas from emission source P1 at a nitric acid plant situated in the industrial zone of Devnya for the period 2006 – 2012 is indicated in Table 1. Data analysis indicates that for the entire monitoring period average monthly NO<sub>x</sub> concentrations in the tail gas from emission source P1 do not exceed ELV. Despite that some differences are observed. At the beginning of the research 2006 – 2008 NO<sub>x</sub> concentrations are quite higher and vary between 300 and 450 mg/m<sup>3</sup>. The highest value for this period is registered in August 2006 – 455.93 mg/m<sup>3</sup>. Quite high concentrations are registered through the entire 2007 – from 373.63 mg/m<sup>3</sup> to 440.01 mg/m<sup>3</sup>.

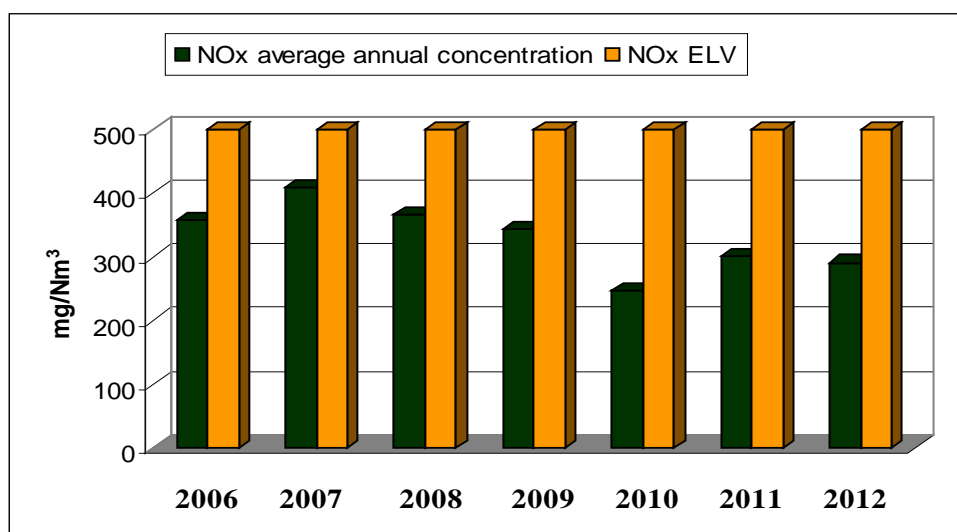
**Table 1. NO<sub>x</sub> concentrations in the tail gas from emission source P1 at nitric acid plant for the period 2006 – 2012, mg/m<sup>3</sup>**

Period	2006	2007	2008	2009	2010	2011	2012
January	316.85	422.57	411.59	322.03	243.03	292.29	291.94
February	317.90	412.47	382.63	344.05	221.83	269.91	295.31
March	322.03	419.33	376.38	358.74	222.10	259.01	295.99
April	343.63	403.37	415.96	372.63	213.16	277.01	*
May	366.20	378.67	348.17	*	218.31	308.32	306.19
June	392.98	440.01	368.99	*	*	346.84	325.89
July	*	*	450.21	*	*	398.62	*
August	455.93	373.63	374.32	382.78	242.94	328.65	344.87
September	379.86	402.70	332.89	380.14	283.77	330.24	263.99
October	339.64	434.70	319.51	333.76	275.35	278.90	253.30
November	329.85	416.42	258.33	311.58	299.74	252.24	313.15
December	388.64	398.26	*	281.27	262.96	273.90	223.39

\* Nitric acid plant is not operating

After 2009 lower concentrations are registered as the lowest value is 213.16 mg/m<sup>3</sup> registered in April 2010 which is 2.14 times lower than the highest value registered for the entire monitoring period. The reason for NO<sub>x</sub> concentration drop during the second half of the monitoring period (after 2009) is the replacement of the secondary decomposing catalyst at the nitric acid plant which took place in November 2009 due to its' reduction potential exhaustion [12].

Figure 1 represents average annual NO<sub>x</sub> concentrations in the tail gas from emission source P1. It is obvious that neither concentration exceeds ELV. The highest average annual concentration is registered in 2007 – 409.28 mg/m<sup>3</sup> while the lowest concentration is 248.32 mg/m<sup>3</sup> registered in 2010. The average annual NO<sub>x</sub> concentration drop follows the replacement of the secondary catalyst.



**Figure 1. Average annual NO<sub>x</sub> concentrations in the tail gas from emission source P1 for the period 2006 - 2012**

### III.2. Quantitative assessment of N<sub>2</sub>O emissions from nitric acid plant

Monitoring data regarding average monthly N<sub>2</sub>O concentrations (mg/m<sup>3</sup>)

in the tail gas from emission source P1 at a nitric acid plant situated in the industrial zone of Devnya for the period 2006 – 2012 is indicated in Table 2.

**Table 2. N<sub>2</sub>O concentrations in the tail gas from emission source P1 at nitric acid plant for the period 2006 – 2012, mg/m<sup>3</sup>**

Period	2006	2007	2008	2009	2010	2011	2012
January	404.00	534.20	732.70	558.30	316.30	448.10	474.90
February	395.00	597.50	773.50	783.90	359.20	474.40	514.50
March	502.00	673.70	867.40	908.60	389.30	550.60	588.70
April	497.50	715.50	1023.50	979.50	454.70	613.80	*
May	563.70	695.00	835.50	*	358.80	663.30	553.20
June	548.70	775.10	700.50	*	*	674.90	491.00
July	*	*	978.30	*	*	373.60	*
August	151.10	406.50	110.50	728.80	223.10	250.50	69.40
September	199.40	525.30	219.60	711.00	343.60	344.40	75.40
October	267.60	609.90	440.80	716.60	418.90	414.70	82.80
November	223.90	624.10	496.30	884.50	471.60	388.50	98.30
December	439.50	707.90	*	448.10	434.60	428.80	105.80

\* Nitric acid plant is not operating

For the monitoring period high temperature catalytic reduction of N<sub>2</sub>O emissions is applied at the nitric acid plant throughout installing a secondary decomposing catalyst which converts N<sub>2</sub>O to oxygen and nitrogen [11]. A research upon the behavior of the catalyst along the production campaign proves that the efficiency of the catalytic reduction depends on the duration of the catalyst's operation and the specific technological conditions of the production process [12]. For that reason higher N<sub>2</sub>O concentrations are registered at the end of the production campaign (months right before the usual plant shut down for platinum gauze pack replacement once a year) compared to those measured at the

start of the campaign (right after the replacement of the platinum gauze pack). A solid trend for higher N<sub>2</sub>O concentrations in the tail gas (over 800 – 900 mg/m<sup>3</sup>) indicates the exhaustion of the secondary catalyst's reduction potential and the necessity of its replacement. The highest average monthly N<sub>2</sub>O concentration for the entire monitoring period is 1023.50 mg/m<sup>3</sup> registered in April 2008 due to nitric acid plant operation at increased capacity at the end of the production campaign. The lowest average monthly N<sub>2</sub>O concentration is 69.4 mg/m<sup>3</sup> registered in August 2012 due to simultaneous replacement of the secondary decomposing catalyst and the platinum gauze pack.

### III.3. Assessment of ambient air quality in Devnya region regarding NO<sub>x</sub>

Monitoring data from automatic monitoring system “Izvorite”

regarding average monthly NO<sub>x</sub> concentrations (µg/m<sup>3</sup>) in the ambient air in Devnya region for the period 2006 – 2012 is indicated in Table 3.

**Table 3. NO<sub>x</sub> concentrations in the ambient air in Devnya region for the period 2006-2012, µg/m<sup>3</sup>**

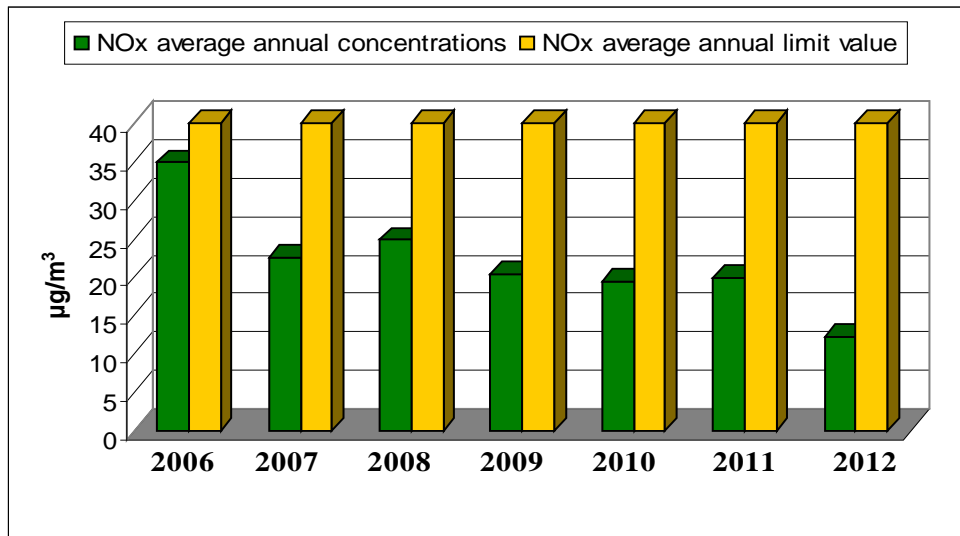
Period	2006	2007	2008	2009	2010	2011	2012
January	20.84	22.01	36.06	*	20.51	27.24	13.30
February	19.77	25.65	33.16	*	20.99	28.09	15.63
March	17.59	26.08	22.27	25.44	18.74	28.13	13.61
April	43.00	28.82	17.92	30.84	18.55	12.86	10.78
May	51.27	27.84	18.70	26.42	14.59	18.97	7.29
June	49.07	26.55	19.16	19.95	13.36	12.51	7.75
July	47.07	32.27	21.22	16.40	13.38	13.22	*
August	47.41	12.62	32.89	14.25	16.96	17.32	10.19
September	32.48	14.89	26.13	14.45	16.21	16.73	10.87
October	24.13	17.08	25.35	10.04	15.93	18.16	10.97
November	30.88	10.53	20.90	24.26	31.21	24.09	10.77
December	*	26.37	23.70	22.33	32.67	21.81	22.28

\* No data available

Analysis of the monitoring data from the automatic monitoring system “Izvorite” indicates that the average monthly NO<sub>x</sub> concentrations are higher at the beginning of the research 2006 – 2008 as well and the highest value is registered in May 2006 – 51.27 µg/m<sup>3</sup>. In 2008 NO<sub>x</sub> concentrations scale up to 36.06 µg/m<sup>3</sup>. After 2009 NO<sub>x</sub> concentrations gradually decrease and in 2012 they vary between 7.29 µg/m<sup>3</sup> and 22.28 µg/m<sup>3</sup>. The NO<sub>x</sub> concentrations drop during the second half of the monitoring period is caused by the significant reduction of NO<sub>x</sub> and N<sub>2</sub>O emissions from the nitric acid plant after the replacement

of the secondary decomposing catalyst.

Results represented on Figure 2 indicate that neither average annual NO<sub>x</sub> concentration for the entire monitoring period exceeds the annual average limit value for human health protection 40 µg/m<sup>3</sup> [9]. A comparable trend is observed regarding the NO<sub>x</sub> pollution of the ambient air - average annual NO<sub>x</sub> concentrations significantly decrease after the replacement of the secondary catalyst at the nitric acid plant as the lowest value for the entire monitoring period is 12.13 µg/m<sup>3</sup> registered in 2012.

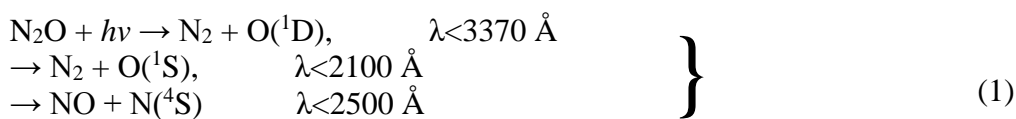


**Figure 2. Average annual NOx concentrations in the ambient air in Devnya region for the period 2006 - 2012**

### III.4. Assessment of NOx emissions upon ambient air quality in Devnya region

NOx emissions mainly contain NO which consequently oxidizes to NO<sub>2</sub> with the participation of ozone and other oxidizers. At oxidizing process unsaturated hydrocarbons and carbon monoxide CO take part as these air pollutants are available in the

environment. Spatial distribution of NO<sub>2</sub> depends mainly on chemical reactions of NO transformation to NO<sub>2</sub>, photochemical equilibrium with ozone and other oxidizing agents in the atmosphere [13]. Nitrous oxide N<sub>2</sub>O is destroyed in the atmosphere by photo dissociation as follows – Equation (1) [14]:



Additional destruction occurs chemically as follows - Equation (2):



It is indicated that N<sub>2</sub>O destruction in the atmosphere leads mainly to formation of NO, nitrogen and oxygen. The content of NO as a product of N<sub>2</sub>O destruction affects the

ambient air quality in addition to other NOx emission sources.

Monitoring data analysis indicates that certain dependence between NOx emission levels from an industrial

source and NO<sub>x</sub> concentration in the More detailed information about the correlation between NO<sub>x</sub> emissions and NO<sub>x</sub> concentrations in the atmosphere can be provided by mathematical modeling the distribution of polluting emissions from emission source P1 in the ground atmospheric layer of Devnya region. Modeling process should take into consideration N<sub>2</sub>O diffusion,

#### IV. Conclusions

In regions where NO<sub>x</sub> and N<sub>2</sub>O emission sources are available ambient air quality regarding NO<sub>x</sub> (NO and NO<sub>2</sub>) content depends on the emission levels and the specific climate and topographic conditions in the region as they define the parameters of pollutants' distribution in the ground atmospheric layer along with the processes of photo dissociation, chemical destruction and diffusion. Nitrous oxide N<sub>2</sub>O is a

ambient air of the region is observed. photo dissociation and chemical destruction in the atmosphere as well as cumulative effect of the formed reaction products upon ambient air quality in the region. Mathematical modeling in order to assess the effect of N<sub>2</sub>O emissions upon ambient air quality is a subject of further scientific research.

greenhouse gas and limiting its' emissions is a significant tool for climate change combat. Monitoring data is not a sufficient source to define the contribution of N<sub>2</sub>O emissions to ambient air quality in the region regarding NO<sub>x</sub> concentrations. That's why mathematical modeling should be done by using appropriate software tool and detailed database containing information about N<sub>2</sub>O emission levels, climate and topographic conditions in the region for the entire monitoring period.

#### References

- [1] The Kyoto Protocol to the United Nations Framework Convention on Climate Change, 1998
- [2] Annual European Union greenhouse gas inventory 1990 – 2010 and inventory report 2012, EEA Technical report № 3 / 2012, Copenhagen, 2012
- [3] Directive 2008/50/EC of the European Parliament and of the Council of 21 May 2008 on ambient air quality and cleaner air for Europe. OJ L 152, 11.6.2008, p. 1-44
- [4] Council Directive 96/62/EC of 27 September 1996 on ambient air

quality assessment and management. OJ L 296, 21.11.1996, p. 55-63

[5] Council Directive 1999/30/EC of 22 April 1999 relating to limit values for sulphur dioxide, nitrogen dioxide and oxides of nitrogen, particular matter and lead in ambient air. OJ L 163, 29.6.1999, p. 41-60

[6] European Commission, Integrated Pollution Prevention and Control, Reference Document on BAT for the manufacture of Large Volume Inorganic Chemicals – Ammonia, Acids and Fertilizers, August 2007

- [7] Agency for Energy and the Environment, Reduction of Nitrous Oxide in the Nitric Acid Industry, The Hague, 2001
- [8] Stefanova M. Industrial methods for nitrous oxide emission reduction. Ecological Engineering and Environmental Protection, 2013 (in press)
- [9] Regulation № 12 of 15 July 2010 г. relating to limit values for sulphur dioxide, nitrogen dioxide and oxides of nitrogen, particular matter and lead in ambient air. Governmental newspaper 58/30.7.2010, p. 27-52
- [10] Regulation № 1 of 27 June 2005 relating to limit values for air pollutants emitted in the atmosphere from plants and activities with immobile emission sources. Governmental newspaper 64/05.8.2005, p. 18-37
- [11] Stefanova M., R. Chaturkova. Nitrous oxide catalytic reduction at a nitric acid plant in Bulgaria. Environment Protection Engineering, 2013 (in press)
- [12] Stefanova M., R. Chaturkova. Research of the efficiency of a secondary catalyst for nitrous oxide emission reduction at a nitric acid plant in Bulgaria. Polish Journal of Environmental Studies, 2013 (in press)
- [13] Chaturkova R., V. Bozhilova, I. Petrova. Dynamics of atmospheric air pollution in an ecologically threatened region. Annual Proceeding of International Medical Association Bulgaria, 1999, vol. 5, No 2, p. 254-256
- [14] Bates D. R., P. B. Hayes. Atmospheric nitrous oxide. Planetary Space Sci., 15, 1967, p. 189-197